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Gibbs excess adsorption and absolute adsorption may be considered equal at low pressure¹ for the Gibbs surface excess discussed in Section 4. Here,

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the Gibbs adsorption equation, is directly connected with the notion about the separating surface. In the Gibbs theory of capillarity,

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Gibbs' adsorption isotherm. $=\sum\{\Gamma_{i}\}d\mu_{i}$ The surface excess: $\Gamma_{i}=\frac{n_{i}^{\sigma}}{A}$ \text{ mol m}^{-2}

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